ELSEVIER

Contents lists available at ScienceDirect

Journal of Alloys and Compounds



journal homepage: www.elsevier.com/locate/jallcom

Formation and physical properties of a novel compound Yb₃Pt₂₃Si₁₁

D. Kaczorowski^{a,*}, A. Gribanov^b, S. Safronov^b, P. Rogl^c, Y. Seropegin^b

^a Institute of Low Temperature and Structure Research, Polish Academy of Sciences, P.O. Box 1410, 50-950 Wroclaw 2, Poland
^b Chemistry Department, Moscow State University, Leninskie Gory, GSP-1, 119991 Moscow, Russia

^c Institute of Physical Chemistry, University of Vienna, Währingerstrasse 42, A-1090 Wien, Austria

ARTICLE INFO

Article history: Received 27 December 2010 Received in revised form 30 June 2011 Accepted 1 July 2011 Available online 12 July 2011

Keywords: Ytterbium intermetallics Crystal structure determination Magnetic properties

1. Introduction

Since many years Ce, Yb, and U-based intemetallics have attracted much research interest owing to their intriguing physical properties such as valence fluctuations, Kondo-lattice behavior, heavy-fermion states, non-Fermi liquid features, unconventional superconductivity (see for instance [1,2]). Recently, in the course of our systematic study of the Yb–Pt–Si system, we identified a novel intermetallic compound Yb₃Pt₂₃Si₁₁. In this paper we report on the crystal structure of the new phase as well as on its magnetic, electrical and thermodynamic behavior at low temperatures. The structural and physical properties of Yb₃Pt₂₃Si₁₁ [3], Ce₃Pt₂₃Si₁₁ [3–6], Ce₃Pt₂₃Ge₁₁ [7,8], and U₃Pt₂₃Si₁₁ [9,10].

2. Experimental details

A polycrystalline sample of Yb₃Pt₂₃Si₁₁ was prepared by melting the elements in an arc furnace on a water-cooled cooper hearth under purified argon atmosphere. Platinum and silicon were taken in stoichiometric amounts, while some excess (~3 mass%) of ytterbium was added due to highly volatile nature of this metal at high temperatures. The button was melted several times, and after each melting the weight losses were compensated by adding the appropriate amount of ytterbium metal (again with ~3 mass% excess). After the final melting, the weight of the sample was equal to that calculated for the ideal composition. No further heat treatment was applied.

The product was characterized by powder X-ray diffraction on a STOE STADI P transmission diffractometer equipped with a linear position sensitive detector. CuK α_1 radiation and Si external standard ($a_{Si} = 0.54307$ nm) were used. The lat-

ABSTRACT

The crystal structure of the novel compound $Yb_3Pt_{23}Si_{11}$ has been determined from powder X-ray diffraction data to be isotypic with cubic $Ce_3Pt_{23}Si_{11}$ (space group $Fm\overline{3}m$; a = 1.68052(5) nm). Magnetic susceptibility, electrical resistivity and heat capacity measurements, performed down to 0.4 K, revealed that $Yb_3Pt_{23}Si_{11}$ is a moderately enhanced paramagnetic compound with fairly unstable 4*f* electronic shell.

© 2011 Elsevier B.V. All rights reserved.

tice parameter was calculated using the STOE WinXPow program [11]. Structure refinement was performed using the Rietveld method implemented in the FULLPROF program [12,13]. Atom parameters were standardized with the aid of the program STRUCTURE TIDY [14].

Magnetic measurements were carried out in the temperature range 1.71–400 K and in magnetic fields up to 5T using a Quantum Design MPMS-5 SQUID magnetometer. The electrical resistivity was measured over the range 4.2–300 K employing a conventional four-point dc technique and a home-made experimental setup. The heat capacity was studied from 6 K down to 0.4 K and in magnetic fields up to 5T using a relaxation method implemented in a Quantum Design PPMS-7 platform.

3. Results and discussion

3.1. Crystal structure

The X-ray diffraction pattern of $Yb_3Pt_{23}Si_{11}$ (see Fig. 1) was properly described assuming a cubic structure of the Ce₃Pt₂₃Si₁₁ type (space group $Fm\overline{3}m, Z=8$). The refined lattice parameter was a = 1.68052(5) nm. The main crystallographic data are summarized in Table 1. Table 2 gives the atomic positions and the isotropic thermal displacement parameters, while Table 3 lists the shortest interatomic distances.

The unit cell of $Yb_3Pt_{23}Si_{11}$ is shown in Fig. 2. The Yb atoms, located at the 24*d* sites, are coordinated by tetragonal prisms build of eight Pt atoms. The Pt atoms occupy four inequivalent positions in the unit cell, whereas the Si atoms are distributed over three different sites. The refined crystal structure is fully ordered. For a detailed discussion of the crystallographic features in the Ce₃Pt₂₃Si₁₁ type structure, the interested reader is referred to Ref. [4].

Yb₃Pt₂₃Si₁₁ is a new representative of ternary compounds crystallizing with large face-centered cubic unit cells ($a \approx 1.7$ nm) such

^{*} Corresponding author. Tel.: +48 71 34 350 21; fax: +48 71 34 410 19. *E-mail address*: D.Kaczorowski@int.pan.wroc.pl (D. Kaczorowski).

^{0925-8388/\$ -} see front matter © 2011 Elsevier B.V. All rights reserved. doi:10.1016/j.jallcom.2011.07.003

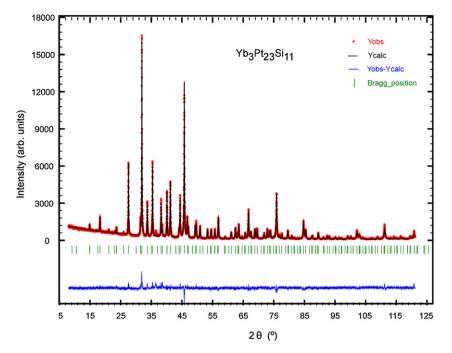
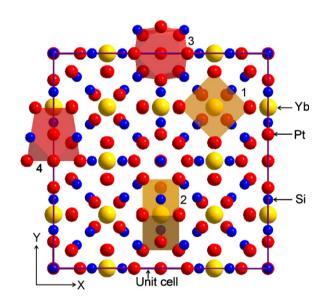


Fig. 1. X-ray diffraction pattern of Yb₃Pt₂₃Si₁₁ with the results of Rietveld refinement.

Table 1Crystallographic details for Yb₃Pt₂₃Si₁₁.

Composition, EPMA (at.%)	Yb _{8.4} Pt _{62.6} Si _{29.0}
Composition from refinement (at.%)	Yb _{8.1} Pt _{62.2} Si _{29.7}
Formula from refinement	Yb ₃ Pt ₂₃ Si ₁₁
Space group	Fm3m (No 225)
Pearson symbol	cF296
Structure type	Ce ₃ Pt ₂₃ Si ₁₁
Lattice parameter (nm)	1.68052(5)
Two theta range (°)	8-121
Step (°)	0.01
Counting time per step (s)	90
Profile function	Pseudo-Voigt
Reflections measured	239
Number of variables	29
$R_F = \sum_{I_o} F_o - F_c / \sum_{I_o} F_o$ $R_I = \sum_{I_o} I_o - I_c / \sum_{I_o} I_o$	0.067
	0.093
$R_{WP} = \left[\sum_{i}^{n} w_{i} \mid y_{0i} - y_{ci} \mid^{2} / \sum_{i}^{n} w_{i} \mid y_{0i} \mid^{2}\right]^{1/2}$ $R_{P} = \sum_{i}^{n} \left y_{0i} - y_{ci}\right / \sum_{i}^{n} \left y_{0i}\right ^{2}$	0.119
$R_{\rm P} = \sum y_{\rm oi} - y_{\rm ci} / \sum y_{\rm oi} $	0.087
$R_{\rm e} = \left[\overline{\rm N} - \rm P + \rm C / \sum w_i y_{\rm oi}^2\right]^{1/2}$	0.045
$\chi^2 = \left(R_{\rm wP}/R_e\right)^2$	7.0



as the closely related compounds hitherto reported in the literature: Ce₃Pt₂₃Ge₁₁ [7,8], Ce₃Pt₂₃Si₁₁ [3–6], La₃Pt₂₃Si₁₁ [3], and U₃Pt₂₃Si₁₁ [9,10]. Whereas the crystal structure of Ce₃Pt₂₃Ge₁₁ was solved within space group $F\overline{4}$ 3*m* (No. 216), all the other phases crystallize with space group $Fm\overline{3}m$ (No. 225) as for the present case of Yb₃Pt₂₃Si₁₁. It is worth mentioning that attempts to adopt for the latter compound the Ce₃Pt₂₃Ge₁₁ type structure (space group

Fig. 2. Projection of the crystal structure of $Yb_3Pt_{23}Si_{11}$ onto the *XY* plane and coordination polyhedra of the Yb (1 and 2) and Pt₃ (3 and 4).

 $F\overline{4}$ 3*m*) yielded significantly larger residual *R* values with markedly worse intensity fitting. Another remark to be made here is that despite Ce₃Pt₂₃Si₁₁ and U₃Pt₂₃Si₁₁ were reported in the literature as isotypic compounds with identical Wyckoff sequences, the two

Table 2

Atomic coordinates and isotropic thermal displacement parameters for Yb₃Pt₂₃Si₁₁.

Atom site	Wyckoff position	x/a	y/b	z/c	$B_{\rm iso} (10^2{\rm nm}^2)$	Occupation
Yb	24d	0	1/4	1/4	0.83(4)	1.0
Pt ₁	32f	0.08337(8)	0.08337(8)	0.08337(8)	0.89(1)	1.0
Pt ₂	32f	0.30751(7)	0.30751(7)	0.30751(7)	0.89(1)	1.0
Pt ₃	24 <i>e</i>	0.3760(2)	0	0	0.89(1)	1.0
Pt ₄	96k	0.08520(5)	0.08520(5)	0.25492(8)	0.89(1)	1.0
Si ₁	24e	0.181(1)	0	0	1.7(1)	1.0
Si ₂	32f	0.1697(5)	0.1697(5)	0.1697(5)	1.7(1)	1.0
Si ₃	32f	0.3902(5)	0.3902(5)	0.3902(5)	1.7(1)	1.0

Atom1	Multiplicity	Atom2	Distance (nm)
Yb	8×	Pt ₄	0.3118
Pt ₁	$1 \times$	Si ₂	0.2513
	3×	Si ₁	0.2567
	3×	Pt_1	0.2802
	3×	Pt ₄	0.2883
Pt ₂	3×	Si ₂	0.2378
	$1 \times$	Si ₃	0.2408
	3×	Pt ₄	0.2698
	3×	Pt ₂	0.2733
Pt ₃	$4 \times$	Si ₃	0.2619
	$4 \times$	Pt ₄	0.2870
	$4 \times$	Pt ₃	0.2948
Pt ₄	$1 \times$	Si ₃	0.2348
	$1 \times$	Si ₁	0.2380
	$1 \times$	Si ₂	0.2466
	$1 \times$	Pt ₂	0.2698
	2×	Pt_4	0.2863
	$1 \times$	Pt ₃	0.2870
	$1 \times$	Pt ₁	0.2883
Si ₁	$4 \times$	Pt ₄	0.2380
	$4 \times$	Pt ₁	0.2567
Si ₂	3×	Pt ₂	0.2378
	3×	Pt ₄	0.2466
	$1 \times$	Pt ₁	0.2513
Si ₃	3×	Pt ₄	0.2348
	1×	Pt ₂	0.2408
	3×	Pt ₃	0.2619

Table 3Main interatomic distances in Yb₃Pt₂₃Si₁₁ (eds < 0.0005 nm)</td>

silicides seem to be occupation variants. While in Ce₃Pt₂₃Si₁₁, the sites 24d (0,1/4,1/4) and 24e (x,0,0) are occupied by Ce and Pt atoms [4], a reverse distribution of atoms was claimed for U₃Pt₂₃Si₁₁ [10]. Fig. 2 demonstrates differently oriented coordination polyhedra for atoms in the site 24d (0,1/4,1/4) (1 and 2) and for atoms in the site 24e (x,0,0) (3 and 4).

3.2. Physical properties

As shown in Fig. 3, the magnetic susceptibility of $Yb_3Pt_{23}Si_{11}$ is nearly temperature independent over an extended temperature range, yet rapidly increases below about 50 K. The observed behavior can be described in the entire temperature range in terms of the so-called modified Curie–Weiss

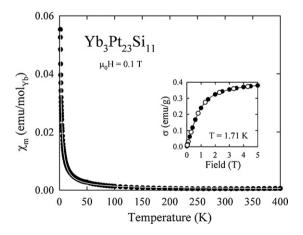


Fig. 3. Temperature dependence of the molar magnetic susceptibility of $Yb_3Pt_{23}Si_{11}$ measured in a field of 0.1 T. The solid line represents the least-squares fit discussed in the text. Inset: field variation of the magnetization in $Yb_3Pt_{23}Si_{11}$ measured at 1.71 K with increasing (full circles) and decreasing (open circles) magnetic field strength.

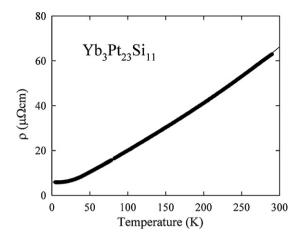


Fig. 4. Temperature dependence of the electrical resistivity of $Yb_3Pt_{23}Si_{11}$. The solid line represents the least-squares fit discussed in the text.

formula $\chi(T) = \chi_0 + C/T - \theta_p$, where χ_0 is a sum of temperature independent contributions, C is a Curie constant, and $\theta_{\rm p}$ stands for the paramagnetic Curie temperature. The leastsquares fit parameters are $\chi_0 = 3.42(4) \times 10^{-4}$ emu/(mol Yb-atom), C = 0.1026(2) (emu K)/(mol Yb-atom) and $\theta_p = -0.2(2)$ K. The small value of the parameter C yields the effective magnetic moment of only $0.9 \mu_B$ per Yb-atom, which is much smaller than the theoretical value calculated within the Russell-Saunders coupling scenario for a stable Yb³⁺ ions (μ_{eff} = 4.54 μ_B). Frequently, Yb-based intermetallics form with divalent Yb ions that have nonmagnetic $4f^{14}$ electronic configuration. In such a case the intrinsic behavior of a given compound is represented by a Pauli-like term χ_0 , while a Curie-like upturn in the magnetic susceptibility is attributed to magnetic impurities (usually some ytterbium oxides or/and hydroxides with Yb³⁺ ions, accumulated on grain boundaries). For the studied sample of Yb₃Pt₂₃Si₁₁ the latter scenario would imply the presence of as much as about 12% impurity atoms per formula unit. This very large amount of foreign phase can be ruled out based on the results of the X-ray diffraction. Therefore, the reduced value of $\mu_{\rm eff}$ and the enhanced value of χ_0 , derived from the $\chi(T)$ data, can be considered as fingerprints of unstable 4f shell in the compound studied. The strongly intermediate-valence (IV) character of Yb₃Pt₂₃Si₁₁ markedly contrasts with the Curie–Weiss paramagnetism of Ce₃Pt₂₃Si₁₁, occurring due to the presence of stable Ce³⁺ ions [3], which eventually give rise to a ferromagnetic phase transition at $T_{\rm C}$ = 0.44 K [6].

The non-magnetic IV nature of Yb₃Pt₂₃Si₁₁ is reflected also in the metallic-like temperature dependence of the electrical resistivity (see Fig. 4). The overall behavior of $\rho(T)$ can be represented by the function $\rho(T) = \rho_0 + 4RT(T/\Theta_R)^4 \int_0^{\Theta_R/T} (x^5 dx)/(e^x - 1)(1 - 1)(1 - 1)(1 - 1)(1 - 1)(1 - 1)(1 - 1))(1 - 1)(1$ e^{-x}) + KT^3 , known as a Bloch-Grüneissen-Mott (BGM) formula [15]. In this expression, ρ_0 represents scattering the conduction electrons on static defects in the crystal lattice, the second term accounts for electron-phonon scattering processes (Θ_R is considered as a measure of the Debye temperature), whereas the T^3 term is due to s-d interband scattering. Fitting the BGM formula to the experimental data of Yb₃Pt₂₃Si₁₁ one obtains the parameters $\rho_0 = 5.9(1) \,\mu\Omega$ cm, $R = 0.166(5) \,\mu\Omega$ cm/K, $\Theta_R = 185(4)$ K, and $K = 4.25(2) \times 10^{-7} \,\mu\Omega \,\mathrm{cm/K^3}$. The small value of the residual resistivity, together with a relatively large residual resistivity ratio $RRR \approx 11$ (RRR is the ratio of the resistivity measured at room temperature and ρ_0), indicates a fairly high quality of the polycrystalline sample measured. The derived value of Θ_R is remarkably smaller than the Debye temperature Θ_D = 266 K, reported for the isostructural compound La₃Pt₂₃Si₁₁ (obtained from heat capacity data) [3]. This apparent discrepancy may be attributed to electronic

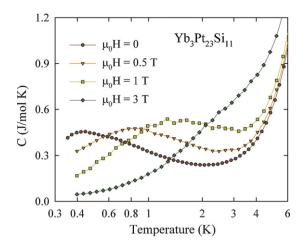


Fig. 5. Temperature dependencies of the low-temperature specific heat of Yb₃Pt₂₃Si₁₁ (note a logarithmic temperature scale) measured in a few different applied magnetic fields. The solid lines serve as a guide to the eye.

correlations, associated with the presumed valence fluctuations in Yb₃Pt₂₃Si₁₁ [16]. Worth noting is also the rather large value of *K*, directly related to the observed curvature in $\rho(T)$, that signifies the importance of scattering the conduction electrons via Mott-type processes.

The low-temperature specific heat data of Yb₃Pt₂₃Si₁₁ are displayed in Fig. 5. The most significant feature is a broad maximum in the C(T) curve that forms below a minimum near 2 K. This behavior seems to correlate with the low-temperature tail in the magnetic susceptibility curve (compare Fig. 3), and hence it likely arises due to unstable 4*f* shell. Applying the afore-made rough estimate that the electronic ground state in Yb₃Pt₂₃Si₁₁ bears in its nature about 12% (per formula unit) of the $4f^{13}$ configuration, one may attempt to qualitatively interpret the maximum in C(T) as a Kondo peak occurring in a system characterized by the total angular momentum I = 7/2. Indeed, the overall shape of the anomaly and the magnitude of the specific heat at the maximum agree reasonably well with the theoretical prediction [17]. Another indication of the Kondo nature of the maximum in C(T) comes from its behavior in applied magnetic field: as may be inferred from Fig. 4, with increasing the field strength one observes a gradual shift of the maximum to higher temperatures with modest increase in its amplitude. The large Kondo contribution yields the enhanced specific heat at the lowest temperatures. Upon suppression of the maximum in C(T)

by applying a magnetic field of 3 T, the *C*/*T* ratio shows a clear tendency for saturation at temperatures $T \rightarrow 0$ (not shown) yielding the electronic Sommerfeld coefficient γ of about 110 mJ/(mol K²). The latter value may be compared with $\gamma = 19$ mJ/(mol K²) reported for the compound La₃Pt₂₃Si₁₁ [3]. This large difference in the electronic contribution to the heat capacity of the two nonmagnetic isostructural phases indicates prominent enhancement in the density of states at the Fermi level, being directly related to the unstable character of the 4*f* states in Yb₃Pt₂₃Si₁₁.

Acknowledgments

This work was supported by the Russian Foundation of Basic Research (project No. 11-03-01191-a) and the Austrian-Polish Scientific-Technical Exchange Program (project PL06/2009).

References

- [1] H.R. Ott, Phys. B: Condens. Matter 378-380 (2006) 1-6.
- [2] M. Brian Maple, R. Baumbach, N. Butch, J. Hamlin, M. Janoschek, J. Low Temp. Phys. 161 (2010) 4–54.
- [3] D.C. Kundaliya, S.K. Malik, Phys. Rev. B 67 (2003) 1-3, 132411.
- [4] A.I. Tursina, A.V. Gribanov, Y.D. Seropegin, K.V. Kuyukov, O.I. Bodak, J. Alloys Compd. 347 (2002) 121-123.
- [5] A. Gribanov, A. Grytsiv, E. Royanian, P. Rogl, E. Bauer, G. Giester, Y. Seropegin, J. Solid State Chem. 181 (2008) 2964–2975.
- [6] C. Opagiste, C. Paulsen, E. Lhotel, P. Rodière, R.M. Galera, P. Bordet, P. Lejay, J. Magn. Magn. Mater. 321 (2009) 613-618.
- [7] A.V. Gribanov, Y.D. Seropegin, O.I. Bodak, V.V. Pavlyuk, L.G. Akselrud, V.N. Nikiforov, A.A. Velikhovski, J. Alloys Compd. 202 (1993) 133–136.
- [8] R. Troc, D. Kaczorowski, T. Cichorek, B. Andraka, R. Pietri, Y.D. Seropegin, A.V. Gribanov, J. Alloys Compd. 262–263 (1997) 211–214.
- [9] C. Geibel, C. Kämmerer, E. Göring, R. Moog, G. Sparn, R. Henseleit, G. Cordier, S. Horn, F. Steglich, J. Magn. Magn. Mater. 90–91 (1990) 435–437.
- [10] J.N. Chotard, O. Tougait, H. Noël, P. Rogl, A. Zelinskiy, O.I. Bodak, J. Alloys Compd. 407 (2006) 36–43.
- [11] STOE WINXPOW (Version 1.06), Stoe & Cie GmbH, Darmstadt, Germany, 1999.
 [12] J. Rodriguez-Carvajal, Satellite Meeting on Powder Diffraction of the XV Congress of the IUCr, Book of Abstracts, Toulouse, France, 1990, p. 127.
- [13] T. Roisnel, J. Rodriguez-Carvajal, in: Proceedings of the European Powder Diffraction Conference (EPDIC7), Mater. Sci. Forum, Book of Abstracts, 2000, p. 118.
- [14] E. Parthé, L. Gelato, B. Chabot, M. Penzo, K. Cenzual, R. Gladyshevskii, TYPIX Standardized Data and Crystal Chemical Characterization of Inorganic Structure Types, Springer-Verlag, Berlin, Heidelberg, 1994.
- [15] N.F. Mott, H. Jones, The Theory of the Properties of Metals and Alloys, Oxford University Press, 1958.
- [16] M. Giovannini, H. Michor, E. Bauer, G. Hilscher, P. Rogl, R. Ferro, J. Alloys Compd. 280 (1998) 26–38.
- [17] V.T. Rajan, Phys. Rev. Lett. 51 (1983) 308-311.